



## Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and  
subscription information:

<http://www.tandfonline.com/loi/gmcl19>

### Ferromagnetism in a New Structural, Phase of $[\text{Fe}(\text{C}_5\text{Me}_5)_2]$ [TCNQ]

William E. Broderick<sup>a</sup>, Xiaohua Liu<sup>a</sup>, Scott Owens<sup>a</sup>, Paul M.  
Toscano<sup>a</sup>, David M. Eichhorn<sup>b</sup> & Brian M. Hoffman<sup>b</sup>

<sup>a</sup> Department of Chemistry, University at Albany, State University of  
New York, Albany, NY, 12222

<sup>b</sup> Department of Chemistry and Materials Research Center,  
Northwestern University, Evanston, IL, 602

Version of record first published: 24 Sep 2006.

To cite this article: William E. Broderick , Xiaohua Liu , Scott Owens , Paul M. Toscano , David M. Eichhorn & Brian M. Hoffman (1995): Ferromagnetism in a New Structural, Phase of  $[\text{Fe}(\text{C}_5\text{Me}_5)_2]$  [TCNQ], Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals, 273:1, 17-20

To link to this article: <http://dx.doi.org/10.1080/10587259508031836>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## FERROMAGNETISM IN A NEW STRUCTURAL PHASE OF [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>][TCNQ]

WILLIAM E. BRODERICK,\* XIAOHUA LIU, SCOTT OWENS, PAUL M.  
TOSCANO

Department of Chemistry, University at Albany, State University of New York,  
Albany, NY 12222

DAVID M. EICHHORN, BRIAN M. HOFFMAN\*

Department of Chemistry and Materials Research Center, Northwestern  
University, Evanston, IL 60208

### Abstract

The five integrated-stack charge-transfer salts [M(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>][TCNQ] (M=Mn, Cr) and [M(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>][TCNE] (M=Fe, Mn, Cr) are bulk ferromagnets, whereas [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>][TCNQ] was reported to exist as two different phases, a metamagnet and a paramagnet. We now report that with careful control of the crystallization conditions, it is possible to isolate a new structural phase of [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>][TCNQ] which is a bulk 3-D ferromagnet with T<sub>c</sub> = 3.1 K. Thus [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>][TCNQ] is the first example of a single compound giving rise to three distinctly different magnetic phenomena, a finding that will permit an assessment of the subtle structural features that control magnetic ordering in this class of materials.

The search for molecular based magnetic materials<sup>1</sup> gained great impetus from the realization of bulk 3-D ferromagnetism in the integrated-stack charge-transfer (CT) donor-acceptor (D<sup>+</sup>-A<sup>-</sup>) salts containing decamethylmetallocenium cations and organic radical anions (Table 1). This class of ferromagnets was discovered in 1987 with the synthesis of [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNE]<sup>-</sup>, with a critical temperature, T<sub>c</sub> = 4.8K.<sup>3</sup> Subsequent demonstration that the [Mn(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup><sup>5</sup> and [Cr(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup><sup>7</sup> analogues as well as [Mn(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNQ]<sup>-</sup><sup>4</sup> and [Cr(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNQ]<sup>-</sup><sup>6</sup> are also ferromagnets not only demonstrated the generality of ferromagnetic ordering in this class of materials,

TABLE 1. Chronology of magnetically ordered integrated-stack CT salts

Compound	D <sup>+</sup> spin	type	T <sub>C</sub> (K)	Date	Reference
[Fe(C <sub>5</sub> Me <sub>5</sub> ) <sub>2</sub> ][TCNQ]	1/2	metamagnet	T <sub>N</sub> = 2.55	1979	2
[Fe(C <sub>5</sub> Me <sub>5</sub> ) <sub>2</sub> ][TCNE]	1/2	ferromagnet	4.8	1987	3
[Mn(C <sub>5</sub> Me <sub>5</sub> ) <sub>2</sub> ][TCNQ]	1	ferromagnet	6.2	1990	4
[Mn(C <sub>5</sub> Me <sub>5</sub> ) <sub>2</sub> ][TCNE]	1	ferromagnet	8.8	1991	5
[Cr(C <sub>5</sub> Me <sub>5</sub> ) <sub>2</sub> ][TCNQ]	3/2	ferromagnet	3.1	1991	6
[Cr(C <sub>5</sub> Me <sub>5</sub> ) <sub>2</sub> ][TCNE]	3/2	ferromagnet	4.0	1993	7

but also resulted in a better understanding of the intrastack interactions responsible for ferromagnetic coupling.

When examining the group, it appears anomalous that [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNQ]<sup>-</sup>, the first magnetically ordered material of this class, is a metamagnet, where antiferromagnetic *interchain* couplings dominate at low applied fields. This suggested that a study of the binary solid solutions [Fe<sub>x</sub>Mn<sub>1-x</sub>(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNQ]<sup>-</sup> might lead to a better understanding of the relevant interstack interactions. However, for values of *x* ranging from 0.1 to 0.9, no deviation from ferromagnetism was discovered, only a monotonic increase in the T<sub>C</sub> with increasing Mn content.<sup>8</sup> Thus we were led to reinvestigate the pure phase, [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNQ]<sup>-</sup>; we find that, in addition to the previously reported paramagnetic and metamagnetic phases, there exists a third distinct phase - a bulk 3-D ferromagnet.

Bulk [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNQ]<sup>-</sup> is prepared by addition of a THF solution of Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub> to a THF solution of TCNQ. Careful control of the crystallization conditions permits us to obtain either green metamagnetic phase II or purple parallelepipeds of the new ferromagnetic phase III. Phase III is isostructural to [Mn(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNQ]<sup>-</sup> and different from the paramagnetic phase I and metamagnetic phase II of [Fe(C<sub>5</sub>Me<sub>5</sub>)<sub>2</sub>]<sup>+</sup>[TCNQ]<sup>-</sup>. While phase I contains isolated D<sup>+</sup>A<sup>-</sup>A<sup>-</sup>D<sup>+</sup> dimeric units, phases II and III consist of ...D<sup>+</sup>A<sup>-</sup>D<sup>+</sup>A<sup>-</sup>... integrated stacks. However, there are clear structural differences between the latter two phases, the most obvious being the conformation of the C<sub>5</sub>Me<sub>5</sub> rings, which are staggered in phase II and eclipsed in phase III.

The magnetic susceptibility of phases II and III were measured with a Quantum Design MPMS SQUID magnetometer (DC susceptibility) and a LakeShore model 7110 AC Susceptometer (AC susceptibility). Magnetization data for phase III taken at an applied field of 5000 G for T > 50 K can be fit to the Curie-Weiss law,  $\chi = C/(T-\theta)$ ,

with  $C=1.26 \text{ cm}^3 \text{ K mol}^{-1}$  and  $\theta = +3 \text{ K}$ , with the positive  $\theta$  suggesting dominant ferromagnetic interactions. Figure 1 shows the magnetization for phase III at low field (15 G) plotted as  $M$  vs.  $T$  for  $T < 10 \text{ K}$ , and an abrupt ferromagnetic transition is clearly present, corresponding to a critical temperature of 3.1 K. Figure 2 shows the AC

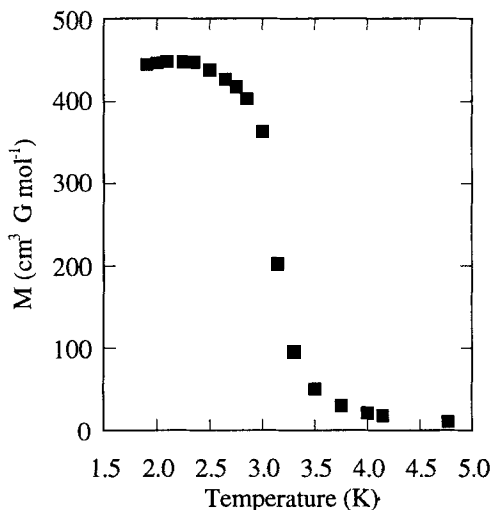


FIGURE 1 Magnetization for phase III of  $[\text{Fe}(\text{C}_5\text{Me}_5)_2]^+[\text{TCNQ}]^-$  at 15 G

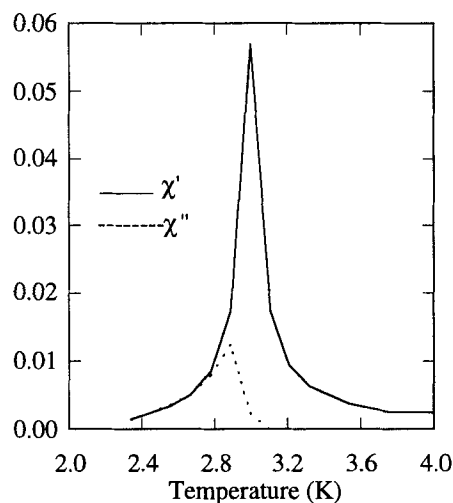


FIGURE 2 AC Susceptibility for phase III of  $[\text{Fe}(\text{C}_5\text{Me}_5)_2]^+[\text{TCNQ}]^-$

susceptibility for phase III. Both the real ( $\chi'$ ) and imaginary ( $\chi''$ ) components show maxima at  $T_c = 3.0 \text{ K}$ , as expected for a bulk ferromagnetic material. The identification of phase III as a ferromagnet was confirmed by investigation of the field dependence of the magnetization. While phase II clearly shows an inflection in  $M$  at  $H \approx 1500 \text{ G}$ , indicating the transition from antiferromagnetic coupling to ferromagnetic coupling that is characteristic of a metamagnet, phase III shows a sharp rise in the magnetization at low fields, reaching a value of  $7500 \text{ cm}^3 \text{ G mol}^{-1}$  at only  $1000 \text{ G}$ , followed by a gradual increase to a value of  $M = 11,600 \text{ cm}^3 \text{ G mol}^{-1}$  at  $50000 \text{ G}$ . This value approaches the saturation magnetization of  $M_{\text{sat}} = 12,100 \text{ cm}^3 \text{ G mol}^{-1}$  expected for a material with  $S_A=S_B=1/2$ ,  $g_A = 2.0$ ,  $g_B = 2.33$ .<sup>9</sup>

Thus the reaction of  $\text{Fe}(\text{C}_5\text{Me}_5)_2$  with TCNQ produces three distinct 1:1 charge-transfer salts: a paramagnet,<sup>10</sup> a metamagnet,<sup>2</sup> and a ferromagnet. To our knowledge,

this is the first example of a single stoichiometry that crystallizes in three forms displaying distinctly different magnetic phenomena. Although the metamagnetic phase II and the ferromagnetic phase III have very similar structures, there are differences both in the juxtaposition of the C<sub>5</sub>Me<sub>5</sub> rings and in the relative interstack approach of the donor and acceptor molecules. Such differences are certainly responsible for the change in sign of the interstack coupling which results in the shift from metamagnetism to ferromagnetism. We are currently engaged in a detailed structural analysis to determine the interchain interactions which control the magnetic behavior.

### ACKNOWLEDGMENTS

This work has been supported in part by Corning, Inc. (W.E.B.), by the Research Foundation of The University at Albany, State University of New York, by the Solid State Chemistry Program of the National Science Foundation through Grant No. DMR-9144513 (B.M.H.) and the Northwestern University Materials Research Center, Grant No. DMR-9120521. We thank D.M. Skee and G.A. Sweeney for technical assistance.

### REFERENCES

1. a) J. S. Miller, A. J. Epstein, W.M. Reiff *Chem. Rev.* , 88, 201 (1988) b) J.S. Miller, A.J. Epstein, W.M. Reiff *Science*, 240, 40 (1988). c) C. Kollmar, O. Kahn *Acc. Chem. Res.*, 26, 259 (1993). (d) C. Kollmar, O. Kahn *J. Chem. Phys.* ,96, 2988 (1992). (e) C. Kollmar, M. Couty, O. Kahn, *J. Am. Chem. Soc.* , 113, 7995 (1991). f) A.L. Buchachenko, *Russ. Chem. Rev.*, 59, 307 (1990).
2. (a) G.A. Candela, L.J. Swartzendruber, J.S. Miller, M. J. Rice *J. Am. Chem. Soc.* , 101, 2755 (1979). (b) J.S. Miller, A.H. Reis, Jr., E. Gebert, J.J. Ritsko, W.R. Salaneck, L. Kovnat, T.W. Cape, R.P. Van Duyne *J. Am. Chem. Soc.* ,101, 7111 (1979).
3. J.S. Miller, J.C. Calabrese, H. Rommelmann, S.R. Chittipeddi, J.H. Zhang, W.M. Reiff, A.J. Epstein, *J. Am. Chem. Soc.* ,109, 769 (1987).
4. W.E. Broderick, J. A. Thompson, E.P. Day, B.M. Hoffman, *Science* ,249, 401 (1990).
5. G.T. Yee, J.M. Manriquez, D.A. Dixon, R.S. McLean, D. M. Groski, R.B. Flippen, K. S. Narayan, A. J. Epstein, J. S. Miller, *Adv. Mater.* , 3, 309 (1991).
6. W. E. Broderick, B. M. Hoffman *J. Am. Chem. Soc.* , 113, 6334 (1991).
7. (a) D. M. Eichhorn, D. C. Skee, W. E. Broderick, B. M. Hoffman, *Inorg. Chem.* , 32, 491 (1993). (b) J. S. Miller, R. S. McLean, C. Vazquez, J. C. Calabrese, F. Zuo, A. J. Epstein *J. Mater. Chem.* , 3, 215 (1993).
8. W.E. Broderick, D.M. Eichhorn, S.M. Owens, B.M. Hoffman; manuscript in preparation.
9. D. M. Duggan; D. N. Hendrickson *Inorg. Chem.* **1975**, 14, 955.
10. A. H. Reis, Jr., L. D. Preston, J. M. Williams, S. W. Peterson, G. A. Candela, L. J. Swartzendruber, J. S. Miller *J. Am. Chem. Soc.* , 101, 2756 (1979).